

Al-Cu-Si (Aluminum-Copper-Silicon)

V. Raghavan

The compilation by [1995Vil] of the experimental data on this ternary system includes a liquidus projection, partial isothermal sections at 955, 850, 750, 650, 600, 500, 400 and 25 °C and twenty vertical sections, mainly from the studies of [1934Mat], [1948Wil], and [1953Phi]. Recently, [2005Pan] made a thermodynamic assessment of this system and presented a computed liquidus projection and several computed vertical sections for Al-rich alloys.

Binary Systems

The Al-Cu phase diagram [1998Sau, Massalski2] depicts a number of intermediate phases: CuAl_2 (C16-type tetragonal, denoted θ), CuAl (monoclinic, η), Cu_5Al_4 (LT) (monoclinic, ζ), ε_2 (NiAs-type hexagonal), ε_1 (bcc), Cu_3Al_2 (rhombohedral, δ), Cu_9Al_4 (HT) (γ_0), Cu_9Al_4 (LT) ($D8_3$ -type cubic, γ_1), and Cu_3Al (bcc, β). In the above, HT = high-temperature and LT = low-temperature. The Al-Si phase diagram [1998Gro] is a simple eutectic system with the eutectic at 577 °C and 12.2 at.% Si. The Cu-Si phase diagram [2000Yan, Massalski2] depicts the following intermediate phases: Cu_3Si (rhombohedral, denoted η), $\text{Cu}_{15}\text{Si}_4$ (cubic, denoted ε), Cu_4Si (tetragonal, denoted δ),

Cu_5Si (β Mn-type cubic, denoted γ), β (high temperature bcc phase), and Cu_7Si (cph, denoted κ).

Computed Ternary Phase Equilibria

The experimental data reviewed by [1992Luk] were used as the base by [2005Pan]. These included results from [1928Gwy], [1934Mat], [1935His], [1953Phi], an [1990Kuz]. The thermochemical data used in the optimization were those of [1985Far] on the heat of melting of the Al-rich eutectic, and of [1997Bel] and [2000Wit] on the heat of mixing of liquid alloys at 1472 °C and 1302 °C respectively. More recently, new results on the solubility of Al and Cu in solid Si in equilibrium with liquid alloys were reported by [2005Yos]. [2005Pan] used the binary descriptions of [1998Sau] (Al-Cu), [1998Gro] (Al-Si) and [2000Yan] (Cu-Si). Ternary interaction parameters for the liquid phase were optimized. The liquidus projection for Al-rich alloys computed by [2005Pan] is shown in Fig. 1. Two transition reactions U_1 and U_2 and the ternary eutectic reaction E occur near the Al corner. Five vertical sections computed 1, 5, and 10 mass % Si, at 10 mass % Cu, and at 80 mass % Al respectively are compared with the experimental data as indicated in Fig. 2-6 [2005Pan]. The agreement is good in all cases. The invariant horizontalals

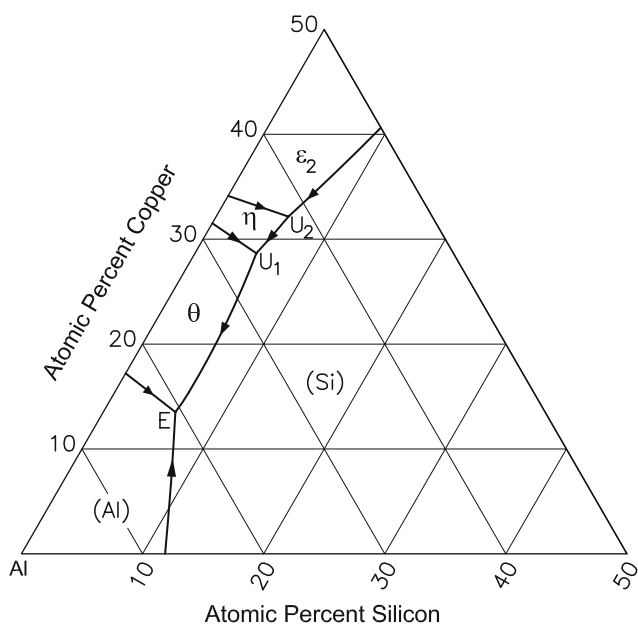


Fig. 1 Al-Cu-Si computed liquidus projection [2005Pan]

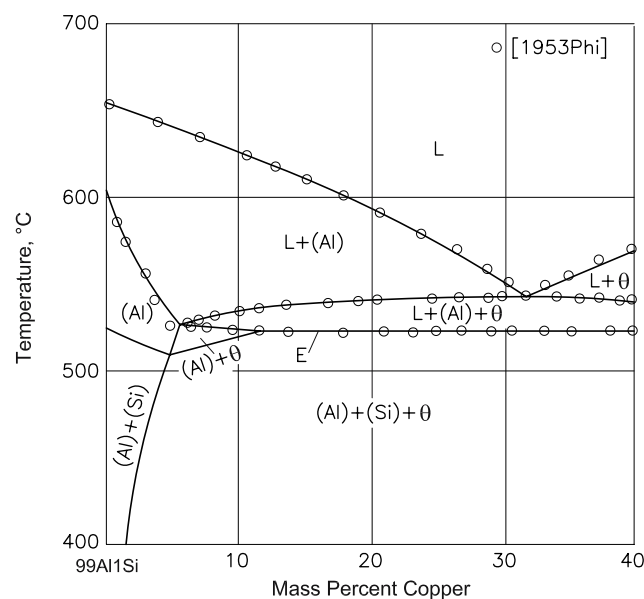


Fig. 2 Al-Cu-Si computed vertical section at 1 mass % Si [2005Pan]

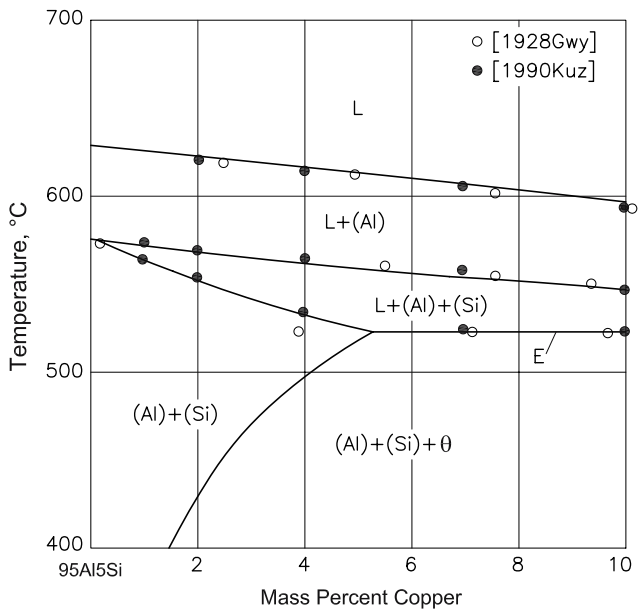


Fig. 3 Al-Cu-Si computed vertical section at 5 mass % Si [2005Pan]

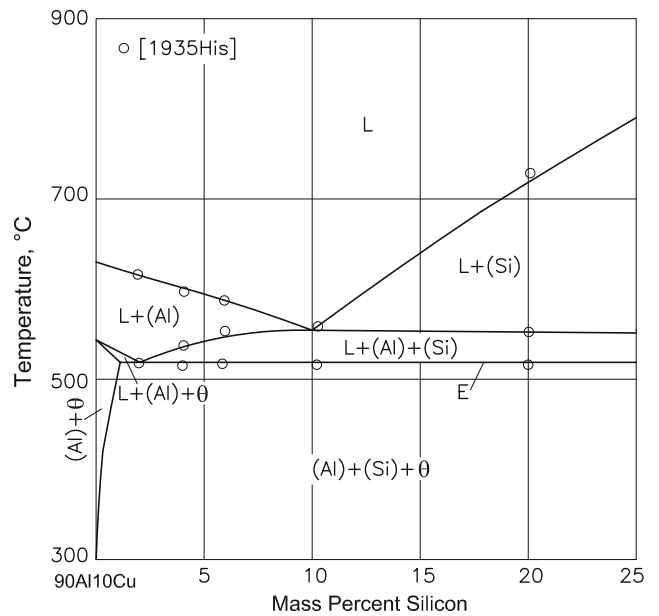


Fig. 5 Al-Cu-Si computed vertical section at 10 mass % Cu [2005Pan]

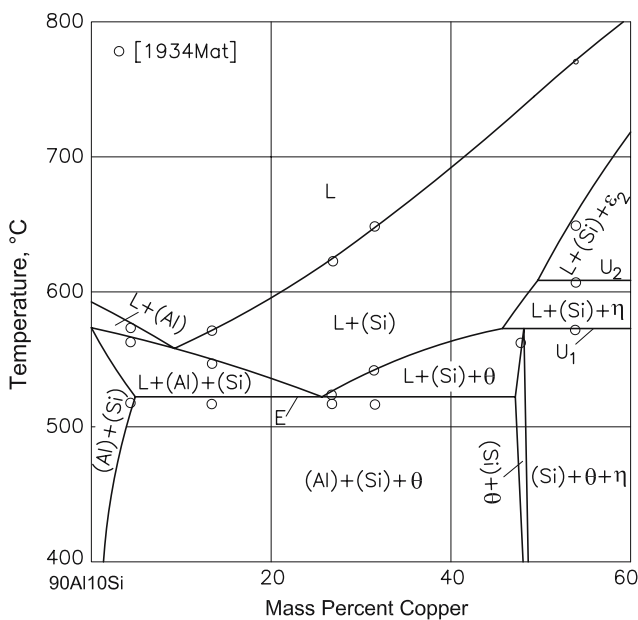


Fig. 4 Al-Cu-Si computed vertical section at 10 mass % Si [2005Pan]

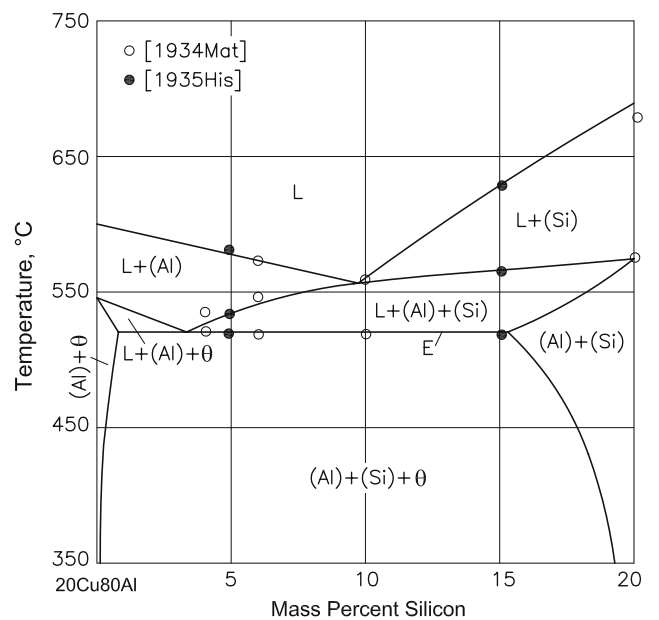


Fig. 6 Al-Cu-Si computed vertical section at 80 mass % Al [2005Pan]

corresponding to E in Fig. 1 is seen in all sections (Fig. 2-6). In Fig. 4, the invariant horizontal lines corresponding to U_1 and U_2 are also seen.

A limited number of experimental results by [2005Pan] on the solidification behavior of Al-7 mass % Si-3.5 mass %

Cu alloy were compared with the computed solidification path under equilibrium conditions and with the Scheil solidification model (infinite rate of diffusion in liquid and nil diffusion in solid).

Section II: Phase Diagram Evaluations

References

- 1928Gwy:** A.G.C. Gwyer, H.W.L. Phillips, and L. Mann, The Constitution of the Alloys of Aluminum with Copper, Silicon and Iron, *J. Inst. Metals (London)*, 1928, **40**, p 297-358
- 1934Mat:** K. Matsuyama, Ternary Diagram of the Al-Cu-Si System, *Kinzoku no Kenkyu*, 1934, **11**, p 461-490, in Japanese
- 1935His:** C. Hisatsune, Constitution Diagram of the Copper-Silicon-Aluminum System, *Mem. Coll. Eng. Kyoto Imp. Univ.*, 1935, **9**(1), p 18-47, in Japanese
- 1948Wil:** F.H. Wilson, The Copper-Rich Corner of the Copper-Aluminum-Silicon Diagram, *Trans. AIME*, 1948, **175**, p 262-273
- 1953Phi:** H.W.L. Phillips, The Constitution of Aluminum-Copper-Silicon Alloys, *J. Inst. Metals (London)*, 1953-1954, **82**, p 9-15
- 1985Far:** D. Farkas and C.E. Birchenall, New Eutectic Alloys and Their Heats of Formation, *Metall. Trans. A*, 1985, **16A**, p 323-328
- 1990Kuz:** G.M. Kuznetsov and L.N. Kalkulova, Phase Equilibrium, Equilibrium, Nonequilibrium Crystallization of Cast Alloys of Al-Cu-Si System, *Tsvetn. Metall.*, 1990, (1), p 102-106, in Russian
- 1992Luk:** H.L. Lukas, Aluminum-Copper-Silicon, in *Ternary Alloys*, G. Effenberg and G. Petzow, Eds., Vol. 5, VCH Verlagsgesellschaft, Weinheim Germany, 1992, p 11-21
- 1995Vil:** P. Villars, A. Prince, and H. Okamoto, Al-Cu-Si, *Handbook of Ternary Alloy Phase Diagrams*, Vol. 3, ASM International, Materials Park, OH, 1995, p 3331-3351
- 1997Bel:** E.A. Beloborodova, T.N. Zinevich, N.V. Kotova, D.S. Kanibolotskii, and V.I. Scherbakov, *Protsessy Lit'Ya*, 1997, **6**, p 38-40, in Russian
- 1998Gro:** J. Grobner, H.L. Lukas, and F. Aldinger, *COST 507 - Thermochemical Database for Light Metal Alloys*, Eds., I. Ansara, A.T. Dinsdale, and M.H. Rand, European Comm. EUR 18499 En, Luxembourg, 1998, p 178-181
- 1998Sau:** N. Saunders, *COST 507 - Thermochemical Database for Light Metal Alloys*, Eds., I. Ansara, A.T. Dinsdale, and M.H. Rand, European Comm. EUR 18499 En, Luxembourg, 1998, p 28
- 2000Wit:** W.T. Witusiewicz, I. Arpshofen, H.J. Seifert, and F. Aldinger, Enthalpy of Mixing of Liquid Al-Cu-Si Alloys, *J. Alloys Compd.*, 2000, **297**, p 176-184
- 2000Yan:** X. Yan and Y.A. Chang, A Thermodynamic Analysis of the Cu-Si System, *J. Alloys Compd.*, 2000, **308**, p 221-229
- 2005Pan:** X.M. Pan, C. Lin, J.E. Morral, and H.D. Brody, An Assessment of Thermodynamic Data for the Liquid Phase in the Al-Rich Corner of the Al-Cu-Si System and Its Application to the Solidification of a 319 Alloy, *J. Phase Equilib. Diffus.*, 2005, **26**(3), p 225-233
- 2005Yos:** T. Yoshikawa and K. Morita, Thermodynamics of Solid Silicon Equilibrated with Si-Al-Cu Liquid Alloys, *J. Phys. Chem. Solids*, 2005, **66**(2-4), p 261-265